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# Porous coatings obtained on titanium after IMPULSE – PEO processes

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# **Porous coatings obtained on titanium after IMPULSE – PEO processes**

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**Abstract.** The surface stereometry and chemical compositions of porous coatings obtained during Plasma Electrolytic Oxidation, PEO/ (Micro Arc Oxidation, MAO) processes with use of impulse regime with different frequencies were the main goal of discussions presented in that paper. All the coatings were characterized by Scanning Electron Microscope (SEM), Energydispersive X-ray Spectroscopy (EDS), X-ray photoelectron spectroscopy (XPS). It was found out that the frequencies of voltage signals used in PEO process as well as their duty ratios have a big influence on porosity, roughness and chemical composition of fabricated layers. In addition, it was noted that all the coatings' compounds contained  $Ti^{4+}$  and  $Ca^{2+}$  or  $Mg^{2+}$  as well as  $PO_4^3$ , and/or  $HPO_4^2$ , and/or  $H_2PO_4$ , and/or  $P_2O_7^4$ .

### **1. Introduction**

The electrolytes consisting of sulfuric(VI) and phosphoric(V) acids are typically used for electropolishing, high-density electropolishing and electropolishing in the magnetic field to obtain passive nanolayers [1-7] but they can be also used for plasma electrolytic oxidation (PEO). Fabrication of much thicker PEO coatings, in the range of micrometers, was proved to be feasible under current or voltage control with DC, pulsed and AC processes. Light metals such as titanium, niobium, tantalum, magnesium, aluminium, and their alloys can be processed [8-9]. Our previous papers characterized porous coatings enriched with calcium, magnesium, zinc and copper obtained under constant voltage conditions [9-16]. The purpose of this work is to compare two shapes of pulsed voltage controlled PEO treatments for coatings enriched with calcium or magnesium, obtained in electrolytes based on concentrated phosphoric acid.

#### **2. Methods**

All PEO processes were performed under two shapes of voltages, i.e. 50 Hz and 100 Hz. The results were recorded by oscilloscope with the shape characteristics, which are presented in Figure 1. In both

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cases the same peak-to-peak voltages with the value 500  $V_{\text{pp}}$  with different frequency and duty ratio were applied. Titanium samples (CP Grade 2) of sizes  $10 \times 10 \times 2$  mm were treated in electrolyte containing 500 g of calcium nitrate(V) tetrahydrate  $Ca(NO<sub>3</sub>)<sub>2</sub>·4H<sub>2</sub>O$  or magnesium nitrate(V) hexahydrate  $Mg(NO_3)_2 \cdot 6H_2O$  in 1 dm<sup>3</sup> of concentrated (85 wt%) phosphoric(V) acid  $H_3PO_4$  in time of 3 minutes. The methodic related to SEM, EDS and XPS equipment's was described in detail in our previous article [18].



**Figure 1.** Shape of PEO voltage used for IMPULSE-PEO treatment a) 50 Hz, b) 100 Hz (during measurements the voltage divider 100 times was used)

#### **3. Results and discussion**

Two magnifications of porous PEO coatings obtained in IMPULSE PEO process, in order to show differences between two voltage frequencies, and they are presented in Figures 2 and 3. The most noticeable difference is the pore size of surfaces obtained at two regimes (IMPULSE-PEO 50 Hz and IMPULSE-PEO 100 Hz). Pores obtained at frequency of 50 Hz have visibly bigger diameters than those once obtained at the frequency of 100 Hz. The differences between coatings enriched in calcium and magnesium are not so obvious and are to be described with roughness parameters elsewhere. However, a possible explanation of the that phenomena may be the different plasma nature during PEO treatment, what corresponds with the applied frequency, duty ratio, as well as with a complex nature of ions transport in changing electric field. In figure 4, EDS spectra of coatings obtained with IMPULSE-PEO enriched in calcium and magnesium at 50 Hz and 100 Hz are presented. Signals of oxygen, titanium, phosphorus and calcium or magnesium were recorded.

 Quantification of the obtained results, presented in tables 1 and 2, has to be interpreted with the consideration of porosity and the layer finite thickness, what in this case. Here also oxygen quantification should be interpreted as an approximate one. However, calcium- or magnesium-tophosphorus (Ca/P,  $Mg/P$ ) ratios, as the reliable quantitative parameters were calculated. According to the recorded results, coatings obtained at frequency of 100 Hz showed higher titanium peak, lower phosphorus, oxygen and magnesium or calcium once, than for treatment at 50 Hz. In case of calcium enriched coatings, the higher Ca/P ratio for coatings obtained at 50 Hz (Ca/P =  $0.13$ ) was recorded, then at 100 Hz  $(Ca/P = 0.10)$ . Magnesium enriched coatings most likely do not show significant differences in Mg/P value, which is in the range of 0.15-0.16. It is also worthy noticing that in the same mass concentration of nitric salts in phosphoric(V) acid, with the use of any presented voltage regime, magnesium enriched coatings would have higher Mg/P ratio than calcium enriched coatings Ca/P ratio.it practically means that titanium signal may partially come from the titanium substrate.



**Figure 2.** SEM images of PEO coatings obtained with IMPULSE-PEO process with magnification of  $500\times$ 



**Figure 3.** SEM images of PEO coatings obtained with IMPULSE-PEO process with magnification of 5000×

**Table 1.** EDS results of porous coatings obtained at 50 Hz and 100 Hz in IMPULSE-PEO process at the voltage  $500$  Vpp in electrolytes consisting of H<sub>3</sub>PO<sub>4</sub> with Ca(NO)<sup>3</sup>.4H<sub>2</sub>O<sub>4</sub>

the voltage 500 vpp in electrolytes consisting of $\pi_3$ PO <sub>4</sub> with Ca(INO3)? $4\pi$ <sub>2</sub> O					
Frequency	Ca. at $\%$	P. at%	$O. at \%$	Ti, at%	Ca/P
50 Hz					
$100$ Hz					).10

**Table 2.** EDS results of porous coatings obtained at 50 Hz and 100 Hz in IMPULSE-PEO process at the voltage 500 Vpp in electrolytes consisting of  $H_3PO_4$  with  $Mg(NO_3)_2.6H_2O$ 



Chemical composition of 10 nm surface layer of the obtained coatings was characterized by means of XPS and it is presented in Figure 4. Recorded XPS spectra for porous PEO coatings obtained at 100 Hz and 50 Hz in electrolytes containing magnesium or calcium indicate, on the base of P 2s and P 2p, the presence of phosphates in the forms of  $PO_4^3$ ,  $HPO_4^2$ ,  $H_2PO_4$ , or  $P_2O_7^4$ , on the base of Ti 2p, the presence of Ti<sup>4+</sup>. In calcium or magnesium enriched coatings,  $Ca^{2+}$  or Mg<sup>2+</sup> are present, respecively. Detailed information is as follows: for calcium enriched obtained at 50 Hz coatings, peaks are C1 s 284.8 eV, O1 s 532.0 eV, Ti 2p3/2 460.8 eV, Ca p23/2 348.4 eV, P 2s 192.4 eV, P 2p 135.3 eV, for magnesium enriched obtained at 50 Hz, peaks are C1 s 284.8 eV, O1 s 531.9 eV, Ti 2p<sub>3/2</sub> 460.4 eV, Mg KLL 306.9 eV, P 2s 191.8 eV, P 2p 134.6 eV, for calcium enriched obtained at 100 Hz coatings, peaks are C1 s 284.8 eV, O1 s 532.2 eV, Ti 2p3/2 460.7 eV, Ca p23/2 348.3 eV, P 2s 192.1 eV, P 2p 134.8 eV, for magnesium enriched obtained at 100 Hz, peaks are C1 s 284.8 eV, O1 s 532.2 eV, Ti 2p<sub>3/2</sub> 460.6 eV, Mg KLL 307.5 eV, P 2s 192.0 eV, P 2p 134.9 eV.



**Figure 4.** EDS spectra of coatings obtained with IMPULSE-PEO: a) enriched in calcium at 50 Hz, b) enriched in magnesium at 50 Hz, c) enriched in calcium at 100 Hz, d) enriched in magnesium at 100 Hz



**Figure 5.** XPS spectra of coatings obtained with IMPULSE-PEO: a) enriched in calcium at 50 Hz, b) enriched in magnesium 50 Hz, c) enriched in calcium at 100 Hz, d) enriched in magnesium 100 Hz

#### **4. Conclusions**

- Both presented impulse voltage regimes can be applied for use with PEO process in order to obtain porous coatings enriched in calcium or magnesium, in electrolytes based on concentrated phosphoric(V) acid and nitrate(V) salts of calcium or magnesium.
- Morphology observed with the use of SEM differs visually in significant level if one compares coatings obtained at 50 Hz and 100 Hz what indicates different response of PEO forming process considered as black box system with no possibility of reliable explanation with the use of a constitution equation. However neural network system can be employed in the future research for the possible PEO system modeling [19-22].
- $\bullet$  Enrichment in calcium or magnesium measured as Ca/P or Mg/P is higher in these conditions for magnesium containing electrolytes, however in case of calcium containing electrolytes, higher enrichment can be achieved at lower frequency.
- Top 10 nm of coatings, which in case of biomedical and catalysis application is in direct contact with environment, contains different possible phosphates (V) as an anion and as well as cations of calcium, magnesium and titanium (IV).

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